

ENERGY RELEASE (Joules x 10⁻¹³)

<p>A.</p> <p>H¹ + H¹ ----) He² ----) H² + e⁺ + ν</p> <p>H² + H¹ ----) He³ + γ</p> <p>He³ + He³ ----) He⁴ + 2H¹</p> <hr/> <p style="text-align: center;">OR</p> <p>He³ + He⁴ ----) Be⁷ + γ</p> <p>Be⁷ + e⁻ ----) Li⁷ + ν + γ</p> <p>Li⁷ + H¹ ----) 2He⁴</p> <hr/> <p style="text-align: center;">OR</p> <p>Be⁷ + H¹ ----) B⁸ + γ</p> <p>B⁸ ----) Be⁸ + e⁺ + ν</p> <p>Be⁸ ----) 2He⁴</p>	<p>1.91 × 2 = 3.82</p> <p>8.79 × 2 = 17.59</p> <p style="text-align: right;"><u>20.58</u></p> <p style="text-align: right;">41.99 (2 per cent ν-loss)</p> <p style="text-align: right;">2.53</p> <p style="text-align: right;">.08</p> <p style="text-align: right;"><u>27.78</u></p> <p style="text-align: right;">41.09 (4 per cent ν-loss)</p> <p style="text-align: right;">0.22</p> <p style="text-align: right;">12.34</p> <p style="text-align: right;"><u>4.81</u></p> <p style="text-align: right;">30.59 (29 per cent ν-loss)</p>
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4H¹ ----) He⁴ direct would give 42.81 x 10⁻¹³ joules (no ν-loss)

<p>B.</p> <p>→ C¹² + H¹ ----) N¹³ + γ</p> <p style="padding-left: 20px;">N¹³ ----) C¹³ + e⁺ + ν</p> <p style="padding-left: 20px;">C¹³ + H¹ ----) N¹⁴ + γ</p> <p>→ N¹⁴ + H¹ ----) O¹⁵ + γ</p> <p style="padding-left: 20px;">O¹⁵ ----) N¹⁵ + e⁺ + ν</p> <p style="padding-left: 20px;">N¹⁵ + H¹ ----) C¹² + He⁴</p> <hr/> <p style="text-align: center;">OR (1/1000)</p> <p>N¹⁵ + H¹ ----) O¹⁶ + γ</p> <p>O¹⁶ + H¹ ----) F¹⁷ + γ</p> <p style="padding-left: 20px;">E¹⁷ ----) O¹⁷ + e⁺ + ν</p> <p style="padding-left: 20px;">O¹⁷ + H¹ ----) N¹⁴ + He⁴</p>	<p>3.12</p> <p>2.00</p> <p>12.08</p> <p>11.77</p> <p>2.77</p> <p style="text-align: right;"><u>9.55</u></p> <p style="text-align: right;">40.10 (6 per cent ν-loss)</p> <p>19.40</p> <p>0.95</p> <p>2.82</p> <p>1.92</p>
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Key to symbols

H	Hydrogen	C	Carbon	γ	gamma-ray
He	Helium	N	Nitrogen	ν	neutrino
Be	Beryllium	O	Oxygen	e ⁻	electron with - charge
Li	Lithium	F	Fluorine	e ⁺	electron with + charge

The series of reactions labelled A is the 'proton-proton' CHAIN, and will be predominant at temperatures between 5 and 10 million degrees.

The series labelled B is the 'carbon-nitrogen' CYCLE ('cycle because the C¹² is re-generated and acts rather like a chemical catalyst) and will be predominant at higher temperatures; the higher temperatures are required to overcome the more powerful electrostatic repulsions of the reactants in this cycle.